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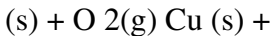
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the steps in writing a balanced chemical equation. Write the skeleton equation for the following reactions:

Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride. CuS



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chemical equation.

answer. A

representation of a
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products on the right,
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skeleton equation.

answer. A chemical
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coefficient. A short, easy way to show a chemical reaction, using symbols. a chemical equation that does not indicate the relative amount. a substance that initiates or accelerates a chemical reaction.

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Product. A new
substance formed in a
chemical reaction. ...

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think enlarged and faster can be undergone by some ways.

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Reactions Work Answer Key

As shown in Figure

11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide

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forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the ...

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Assume there are no side reactions or auxiliary reactions.

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From Eqs. 11.5.9 and 11.5.10, calculate the standard molar internal energy of combustion of n-hexane at (298.15K) . (p) From Eq. 11.5.13, calculate the standard molar enthalpy of combustion of n-hexane at (298.15K) . 11.8

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