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$11 \times N = 14$ ;  $3 \times H = 3$ ;  $1 \times S = 32$ ;  $3 \times O = 16$   
 $\times 3 = 48$ . 2Empirical mass =  $14 + 3 + 32 + 48 =$   
97. 3Divide molar mass by empirical mass:  
 $97.01/97 = 1$ , therefore molecular formula =  
empirical formula. 4© Pearson Education Ltd  
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qualifications~~

Plants can obtain C as  $CO_2$ , N as ammonia or  
nitrite or nitrates, phosphorus as

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phosphates, sulfur as sulfides or sulfates, and so on. In other words, plants obtain these elements as inorganic compounds. 2. A chemical element cannot be broken down to other forms by chemical reactions.

~~Answers — SharpSchool~~

SAMPLE ANSWER: Organisms use carbon compounds to form four types of molecules: lipids, carbohydrates, nucleic acids, and proteins.

SAMPLE ANSWER: A lot of what that happens in an organism is based on chemical reactions.

SAMPLE ANSWER: Enzymes are proteins that speed up chemical reactions that take place

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in cells.

## ~~The Chemistry of Life~~

an element are similar, the symbol will appear to have been derived from the English name. •Examples include Ca for calcium, N for nitrogen, and S for sulfur. Symbols and Formulas. 2.3 Elements and Compounds >. 37. Symbols and Latin Names for Some Elements.

## ~~Chapter 2~~

Compounds Chemical compounds tend to form so that each atom, by gaining, losing, or sharing electrons, has an octet of electrons



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surrounding the atom. Charge of an Ion.  
Cation-> + (positive) Page 1/5. Acces PDF  
Chapter 7 Ionic And Metallic Bonding Answer  
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the name of the element ?the word ion or  
cation. the root of the name of the element  
?-ide. Sample answers: cation: sodium ion;  
anion: chloride ion; positive polyatomic ion:  
ammonium ion; negative polyatomic ion:  
sulfate ion. 0132525887\_CHEM\_WKBK\_CH 09.indd  
110 3/24/10 6:06:54 PM. Name Class Date.

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~~Chemical Names and Formulas — AP Biology~~  
Elements and Compounds 1. A(n) is any substance that cannot be broken down into simpler substances. Its smallest unit is the 2. When two or more elements combine chemically, they form a(n). Its smallest unit is usually called a(n) 3. Most chemical reactions within cells could not take place without O Pearson Education, Inc., publishing as Prentice Hall.

~~Guided Reading Study Chemical Compounds in Cells~~

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Section 1: Chemical Compounds in Cells

Objectives: 1. Define elements and compounds.  
2. Explain how water is important to the function of cells. 3. Identify the four main kinds of organic compounds in living things.

~~Section 1: Chemical Compounds in Cells — Ms. Lariviere's ...~~

Chemical Compounds in Cells Understanding  
Main Ideas Fill in the blanks in the table below. Building Vocabulary Match each term with its definition by writing the letter of the correct definition in the blank beside the term. Organic Compounds Type of Compound

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Example Major Roles in Living Things

Carbohydrates 1. \_\_\_\_\_ Help form cell walls  
and

~~Cell Structure and Function Chemical  
Compounds in Cells~~

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Name of compound. Chemical formula. Number of elements in the compound. Names of elements in the compound. magnesium oxide.  $\text{MgO}$ . 2. magnesium, oxygen. potassium oxide.  $\text{K}_2\text{O}$ .  $\text{FeS}$ . iron, sulphur. potassium hydroxide.  $\text{KOH}$ . 3. calcium carbonate.  $\text{CaCO}_3$

~~New Page 2 [www.scieebhop.net]~~

Introductory Chemistry: Concepts and Critical Thinking, 6th Edition © 2011 Pearson Education, Inc. Charles H. Corwin Example Exercise 7.1 Classifying Compounds and Acids. We can refer to the periodic table and classify each compound or solution as

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follows: (a) CaO contains two elements, a metal and nonmetal. Thus, CaO is a . binary ionic compound. (b) SO. 2

## ~~Example Exercise 7.1 Classifying Compounds and Acids~~

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Chapter 9 Chemical Names and Formulas83  
SECTION 9.3 NAMING AND WRITING FORMULAS FOR~~



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MOLECULAR COMPOUNDS (pages 268-270) This section explains the rules for naming and writing formulas for binary

~~Name Date Class CHEMICAL NAMES AND FORMULAS 9~~

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compounds an atom consist of a positively

charged ...

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Paper: 2C Thursday 16 January 2014 -

Afternoon Time: 1 hour You must have: Ruler

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and candidate number.t t Answer all

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