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[Molarity Chemistry If8766 Instructional Fair](#) [Molarity By Dilution Chemistry If8766 own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar. Molarity Chemistry If8766 Answers Pg 68](#) [Bookmark File PDF Chemistry If8766 Molarity Answers By Dilution Chemistry If8766 Answers Page 8/25](#)

[Molarity By Dilution Chemistry If8766](#) where the subscripts "1" and "2" refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in the solution,  $n_1 = n_2$ . Thus, these two equations may be set equal to one another:  $[M_1 \cdot V_1 = M_2 \cdot V_2]$  This relation is commonly referred to as the dilution equation.

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